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Marco Cascella

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta h = 2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g})$ $\Delta h = 137 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$ from the following data
 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ}$

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 $\text{Mg}(\text{s}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{MgO}(\text{s})$

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 cohen 2016 the equations for the reactions in the experiment
 done are as follows 1 naoh s na aq oh aq

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 ability to do work or transfer heat work is the transfer of
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 the process of transfer

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 seufert kelsea floyd convert joules to kj in your final answer
 multiply the mass by the change in temperature and the c
 given trial 1 use hess s law and the following equations and
 Δh values to determine the heat of reaction for the reaction c
 2 h 4

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web thus for reaction 9 2 in which 1 mole of H_3O^+ and 1 mole
 of OH^- combine to form 2 moles of H_2O $\Delta h = 55.8 \text{ kJ}$ as shown note
 that the enthalpy change $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$ is positive if
 heat is absorbed that is if $\Delta h_{\text{products}}$

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 the goal of this lab is to perform experiments and practice
 hess s law by determining enthalpy change in chemical
 reactions in this lab three experiments were conducted in
 order to test hess s law the experiments required observing
 and collecting data on three different reactions

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web objective the purpose of the lab is to combine two
 reactants in the calorimeter and measure the heats of
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