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ions and hydroxide ions acidic solution one which H^+ is
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than $1 \times 10^{-7} \text{ M}$ they are the same as basic solutions

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determine their weakest conjugate acid the conjugate acids
of CH_3NH_2 and S^{2-} are CH_4 NH_3 and HS^- respectively next we
consider that acidity increases with positive charge on the
molecule thus ruling out that S^{2-} is the weakest base

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bases taste salty and are referred to in chemistry as salts
sweet compounds have characteristics of both acids and
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