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notes 323 nuclei and radioactivity physics module 7 atoms and nuclei z explain the growth and decay of radioactivity in a sample z calculate the half life and decay constant of a radioactive substance and z explain the uses of radioactivity in various fields 26 1 the atomic nucleus soon after the discovery of nucleus in an atom by rutherford in 1911
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 quantum physics yet even though it has only 10 15 of the volume of the atom the nucleus contains more than 99 9 of the mass of the atom the nucleus is very small but very massive that was not predicted try to imagine the surprise and disbelief of scientists in 1911 when ernest rutherford discovered this incredible fact

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over 3 000 different kinds of atomic nuclei have been
discovered in elements that span the periodic table yet each
one has unique properties some may exist forever others
transform within a fraction of a second some can power
satellites as they travel to the outer reaches of the solar
system others have special properties that allow them to
treat different forms of cancer

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radioactivity some nuclides have an atomic nucleus that
decays spontaneously without any external influence known
as radionuclides they transform themselves into other nuclei
ionising radiation is emitted during this process this is
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it some future development of atomic theory perhaps a
modification of the thomson model would most likely yield a
causal explanation of radioactivity in terms of the internal
structure of the atom

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nuclei and their constituents and interactions in addition to
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should not be confused with atomic physics which studies the
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