

145 Meiers Road Indooroopilly

J Elliott

Question #5b826 - Socratic $[\text{Br}^-]_{\text{eq}} = 0.0108 \text{ M}$ $[\text{Cl}^-]_{\text{eq}} = 0.0108 \text{ M}$ $[\text{BrCl}]_{\text{eq}} = 0.0284 \text{ M}$ What we're really doing here is that we're given an initial state with $[\text{BrCl}]_i = 0.050 \text{ M}$, and we're given K_{eq} , which is defined for a final state we define as equilibrium, when the rate of the forward reaction is equal to that of the reverse reaction.

Concentrations are a matter of "what did we start with ... **Question #88b4c - Socratic** The chemical equation and the molar masses are $\text{M r:mmml286.14mmmmmmmmmmmmmmmm18.02 mmmNa}_2\text{CO}_3 \cdot 10\text{H}_2\text{O} \rightarrow \text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O} + 9\text{H}_2\text{O}$ Step 1.

Calculate the moles of $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$. Moles of $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O} = 100.0\text{g Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O} \times 1 \text{ mol Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$

$286.14\text{g Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O} = 0.3495 \text{ mol Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ Step 2. Calculate the moles of water ... **Question #48bcb +**

Example - Socratic From this, the number of unpaired electrons seems to be: Number of unpaired electrons = $S |\text{ms}| =$

$2.145 \frac{1}{2} = 4.29 \approx 4$ So it looks like treating $\mu\text{S} + \text{L} \approx \mu\text{S}$ for Fe^{2+} is OK to determine the number of unpaired electrons. For

Fe^{3+} it should be that $\mu\text{S} + \text{L} = \mu\text{S}$ perfectly, since the 3d is ... **Question #509cd - Socratic** 196 ft.^2 Refer to figures below

From the given data we have $D_1 = 6 \text{ ft.} \Rightarrow r_1 = 3 \text{ ft.}$ $D_2 = 4 \text{ ft.} \Rightarrow r_2 = 2 \text{ ft.}$ $h = 12 \text{ ft.}$ As we can see in Fig. (a) $r_2/(H-h) = r_1/H$

$2/(H-12) = 3/H \Rightarrow 2H = 3H - 36 \Rightarrow H = 36$ Since $s = \sqrt{2} \cdot r$ (see Fig.(b)) $s_1 = 3\sqrt{2}$ $s_2 = 2\sqrt{2}$ Area of bases

$S_{(b1)} = s_1^2 = (3\sqrt{2})^2 = 18$ $S_{(b2)} = s_2^2 = (2\sqrt{2})^2 = 8$ Side of the slanted trapezoid ... How do you add

$(-8-9i) + (3-6i)$ in trigonometric form? | Socratic Jan 25, 2018 · So, for $\#z_1 = -8-9i\#$, we can find that $\#r_1 = \sqrt{(-8)^2 +$

$(-9)^2} = \sqrt{145}\#$ $\#\theta_1 = \arctan((-9)/(-8))\#$ However, $\#-8-9i\#$ is in quadrant 3, but $\#\theta_1\#$ is in quadrant 1, so we

need to add $\#\pi\#$ $\#\theta_1 = \arctan((-9)/(-8)) + \pi \sim 3.9857\#$ $\#z_1 \sim \sqrt{145} (\cos(3.9857) + i \sin(3.9857))\#$ For

$\#z_2 = 3-6i\#$, $\#r_2 = \sqrt{3^2 + (-6)^2} = \sqrt{45} = 3\sqrt{5}\#$ **Question #5d6cd - Socratic** Mar 17, 2016 · Notice that the

reaction ended up producing 145 g of xenon tetrafluoride. This represents the reaction's actual yield, i.e. what you actually

get by doing the reaction in the lab. **Sans the lone satellite Luna, our planet Earth had cleared** Another (2015 TB 145)

came a little beyond Moon's maximum apogee distance of 405400 km. In view of all these findings, it is reasonable to admit

that the Earth is yet to clear some NEOs like the asteroid (2016 RB1) that had come as close as 40000 km to the Earth.

Question #a6cce - Socratic The wavelength is $\lambda = 1.03 \times 10^{-7} \text{ m}$ Detailed solution below... There is a variety of ways

to calculate this, but I will use the method that I think best suits a chemistry approach to the question. The energy of each

stationary state of hydrogen is given by $E_n = -1311/n^2$ where I use a value for the numerator that has units of kJ per mole.

(This is the equivalent of 13.6 eV if you ... If the measures of a triangle are 90°, 55°, and 3x - Socratic $\Rightarrow x = 10$ The measures (angles) of a triangle sum to 180°. $\Rightarrow 90 + 55 + (3x+5) = 180 \Rightarrow 145 + (3x+5) = 180 \Rightarrow 3x + 5 = 35 \Rightarrow 3x = 30 \Rightarrow x = 10$ *Question #48890 - Socratic* 145 degrees Two angles are supplementary when they add up to 180 degrees. Therefore, if an angle, theta, is supplementary to 35 degrees, then $\theta + 35 = 180$ $\theta + 35 = 180$ $\theta = 145$ That is our answer!

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