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chemical change because after its combustion a new substance is formed and cannot be turned back into

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web q5 identify the reducing agent in the following reactions  
a  $4\text{nh}_3 + 5\text{o}_2 \rightarrow 4\text{no} + 6\text{h}_2\text{o}$  b  $\text{h}_2 + \text{o}_2 \rightarrow 2\text{hf}$  c  $\text{fe}_2\text{o}_3 + 3\text{co} \rightarrow 2\text{fe} + 3\text{co}_2$   
d  $2\text{h}_2 + \text{o}_2 \rightarrow 2\text{h}_2\text{o}$  answer a here ammonia  $\text{nh}_3$  is the reducing agent b here water  $\text{h}_2\text{o}$  is the reducing agent