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temp is 20 °C and that potassium nitrate is selected question
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solvent observe click OK click add 20 g and observe the
potassium nitrate being mixed into the solution

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amounts of a chemical to a beaker of water to create a
solution observe that the chemical dissolves in the water at
first and then measure the concentration of the solution at
the saturation point

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maximum concentration of the solute what is the solubility of
potassium nitrate in 20 °C water 3 experiment click reset and
select sodium chloride with the temp still set to 20 °C click OK
add sodium chloride to the beaker until it starts piling up at
the bottom

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of the water c the water is unsaturated d the water is
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