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aqueous ionic compounds or acids

consisting of aqueous ions a and b c

and d when a double replacement

reaction occurs the cations and anions

switch partners resulting in the

formation of two new ionic compounds

ad and cb one of which is in the solid

state

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web complete the changes in

concentrations for each of the following

reactions baso 4 s ba 2 a q so 4 2 a q x

ag 2 so 4 s 2 ag a q so 2 4 a q x al oh 3

s al 3 a q 3 oh a q x pb oh cl s pb 2 a q
oh a q cl a q x

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reactions are usually represented solely by net ionic equations if all products are aqueous a net ionic equation cannot be written because all ions are canceled out as spectator ions therefore no precipitation reaction occurs

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→ ce ca c₂h₃o₂ 2 aq ce 2h₂o l
complete ionic ce ca 2 aq ce 2oh aq ce 2h aq ce 2c₂h₃o₂ aq → ce ca 2 aq ce 2c₂h₃o₂ aq ce 2h₂o l

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