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moles is important in calculating the theoretical yield and how much stuff is reacting mass wise stoichiometry

a key step in balancing chemical equations is correctly identify

web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO and H_2O to form aqueous calcium hydroxide Ca(OH)_2 is it possible to balance the equation if you incorrectly identify the product as CaOH and if so what is the equation

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