

Características De Los Gases Nobles

Clemens Wendtner

Química inorgánica Therald Moeller, 1981-02

Conceptos química inorgánica Bodie Eugene Douglas, John J. Alexander, 1987 Este texto debe ayudar a los estudiantes a integrar sus conocimientos de Química, capacitándolos para aprovechar el caudal de conocimientos adquiridos en cursos de Química independientes. Desde la primera edición de este libro, los avances de la Química inorgánica han sido impresionantes. Para mantener el texto al día se han hecho las adiciones correspondientes y necesarias. Se han utilizado figuras más profusamente que en la primera edición y, asimismo, se han seleccionado cuidadosamente para que resulten más efectivas.

Química: la Ciencia Central Theodore L. Brown, H. Eugene LeMay, Bruce Edward Bursten, Julia R. Burdge, 2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish. Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Química Andoni Garritz, 1998

Fundamentos de la química general e inorgánica Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos- ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dichas- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

Principios de Química. La búsqueda del conocimiento. P. W. Atkins,Loretta Jones,2006

Química. Fundamentos experimentales Robert W. Parry,1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseanza donde se cursan estudios de nivel medio y preuniversitario.Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día.Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

Química 1. Fundamentos ,

Física general,2003 Un clásico entre los manuales de física universitaria, incluye todos los conocimientos que se requieren en física general.Con el objetivo de reforzar los conocimientos teóricos adquiridos en cada tema, se proponen a lo largo de todo el texto un total de 2.100 problemas cuya solución se encuentra en el libro problemas de física de los mismos autores y también publicado por Editorial Tébar.

Química ,2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiométría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Química general Hans Rudolf Christen,1975 El libro Química general, escrito como libro de enseñanza para centros de bachillerato, escuelas de comercio, seminarios y otras instituciones docentes de nivel medio se publicó por primera vez en Suiza, en 1963 y, desde entonces se han efectuado seis ediciones sin cambios notables.Como corresponde a su título, este libro da en primer lugar una visión y conocimiento generales de la materia. Porque no es precisamente un conocimiento detallado y específico de la materia lo que da un valor a la enseanza de la Química, sino un conocimiento del conjunto de leyes fundamentales y de su aplicación al mundo material, de las relaciones principales en el mismo, así como de los métodos (y ilimitaciones!) de la investigación.

Química Paul R. O'Connor,1977 Este libro, es junto con el Manual de laboratorio, fruto directo del programa CHEM publicado también por esta Editorial. Como una de las versiones autorizadas del citado proyecto, esta obra refleja todo el esfuerzo y dedicación del equipo original del proyecto CHEM. El título Química. Experimentos y teorías responde perfectamente bien a la idea básica de este libro; en él se exponen cuidadosamente y además se utilizan a lo largo del mismo, todos los pasos por los cuales transcurre el llamado método científico. Las observaciones experimentales y las medidas dan lugar al desarrollo de los principios teóricos que las unifican y que, más tarde, se utilizan para relacionar e interpretar

diversos fenómenos.

introducción a la química descriptiva ,

Química 2 ,

Materiales industriales. Teoría y aplicaciones Ligia María Vélez Moreno,2008

Química G. William Daub,William S. Seese,1996

El libro de la química (The Chemistry Book) DK,2023-03-28 ¿De qué está hecho el universo? ¿Cómo se crea la materia?

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Ambito Científico Tecnológico I 2025 Francisco Javier Guerra Perlado,Mercedes Sánchez Ruiz,Ascensión Ruano Calvo,

Fundamentos de Química Ralph A. Burns,2003

Química general Ramírez Regalado, Víctor Manuel,2020-05-13 Química general. 2^a. Ed. Esta obra se diseñó con base en los programas de estudio actualizados de la Escuela Nacional Preparatoria de la UNAM. Esta materia abarca un curso anual y básico para el estudio de la química. La obra incluye los contenidos conceptuales, procedimentales y actitudinales que marca la ENP-UNAM. Toda la información está totalmente actualizada y el libro cuenta con una gran variedad de recursos didácticos.

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