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a write the balanced chemical equation
for the reaction that occurs when
solutions of barium chloride and
sodium carbonate are mixed refer to
table 1 on page 437 in chapter 13 for
solubility $\text{BaCl}_2 \text{ aq} + \text{Na}_2\text{CO}_3 \text{ aq} \rightarrow \text{BaCO}_3 \text{ s} + 2\text{NaCl aq}$ b to which of the five basic
types of reactions does this reaction
belong double displacement 10

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section 2 chemical equations a
chemical equation uses symbols to

represent a chemical reaction and
shows the relationship between the
reactants and the products of a
reaction in words carbon dioxide and
water react to form glucose and oxygen
gas CO_2

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trading partners the balanced equation
is $\text{Ce} + \text{NaCN} \rightarrow \text{Ce} + \text{HBr} \rightarrow \text{aq}$
right $\rightarrow \text{Ce} + \text{NaBr} \rightarrow \text{aq}$ right $\text{Ce} + \text{HCN} \rightarrow \text{g}$ right think about your result
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copper sulfur copper i sulfide
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