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How do acid and bases react? + Example - Socratic Mar 22, 2016 · When an acid and a base are placed together, they react to neutralize the acid and base properties, producing a salt (neutralisation). The H (+) in the acid combines with the OH (-) in the base to form water (colourless). The compound formed by the cation of the base and the anion of the acid is called a salt. Example H Cl + N aOH → N aCl + H 2O Hydrochloric acid + ... What are the five most common mineral groups found in rock? Jun 1, 2015 · Halite (NaCl) is perhaps the most common halite mineral. Oxides Oxide minerals contain the oxide O2- or hydroxide OH- ion combined with a metal ion. Aluminium is the third most abundant element in the Earth's crust. Its main ore is bauxite, which contains three minerals, one of which is boehmite, AlO (OH). Question #da59d - Socratic Jan 7, 2015 · M g(OH)2(s) \neq M g2+ (ag) + 2OH – (ag) In order to determine the maximum concentration of M g2+ ions permissible in the N aOH solution before a precipitate will be formed, you'd need the value of the solubility product constant, Ksp. Question #71ce2 - Socratic Nov 9, 2014 · H^+ + OH^--> H 20 when the acid was added to the resulting solution. The H^+ and OH^- react in a 1:1 ratio. This tells us that the number of moles of H^+ used will be equal to the number of OH^- moles in solution. Likewise, 2 moles of lithium produces 2 moles of OH^-. This is also a 1:1 ratio. What are the spectator ions in the reaction between sodium Oct 16, 2016 · The nitrate and the natrium ions. Na 2CO 3(ag) + 2AgNO 3(ag) rarr Ag 2CO 3(s)darr + 2NaNO 3(ag) The net ionic equation is: $2Ag^{(+)} + CO 3^{(2)}$ rarr Ag 2CO 3(s)darr ... What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M + ion. Hydroxide anion, -OH, has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry between ions: Li(s) + H 2O(l) \rightarrow LiOH (ag) + 1 2 H 2(g) $\uparrow \sqcap \sqcap \sqcap$ What is the product of the following reaction? 1)CH 3 OH - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... NH 4^+ +HO^(-) rarr NH 3(ag) + H 2O(l) For methanol, the acid base reaction would proceed ... Could a **buffered solution be made by mixing aqueous ... - Socratic** Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered solution? Calculating the concentration of excess HCL in E - Socratic The acid in excess is then titrated with N aOH (aq) of KNOWN concentration....we can thus get back to the concentration or molar quantity of M (OH)2...as it stands the question (and answer) are hypothetical... C3H7OH+O2 synthesis? - Socratic Apr 19, 2018 · See below When a

substance rapidly reacts with oxygen gas, an alcohol in this case, we call it combustion, so that will be your reaction type and not synthesis When combustion occurs in areas with adequate oxygen, the products are always CO_2 and H_2O 2C_3H_7OH(l)+9O_2(g)->6CO_2(g)+8H_2O(l)

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