

Tabla Periodica Metales Y No Metales

G. William Daub, William S. Seese

Quimica 2 (UdeG) ,

Quimica: la Ciencia Central Theodore L. Brown, H. Eugene LeMay, Bruce Edward Bursten, Julia R. Burdge, 2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish. Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Fundamentos de Quimica Ralph A. Burns, 2003

La tabla periódica Desmond Grosvenor Cooper, 1976 Es importante que el estudiante moderno pueda apreciar y comprender, desde el principio, cómo es el armazón en que está edificada la materia que estudia, en vez de sobrecargado con datos y más datos y, Mr. Cooper, al presentar una monografía concisa y legible sobre la Tabla periódica y su aplicación, ha prestado un valioso servicio a la Química inorgánica. El libro resulta apropiado, idealmente, para estudiantes de los últimos cursos de bachillerato y primer año de Universidad y merece ser recomendado cálidamente.

Quimica 2 ,

Especiación química y física de metales en la materia particulada atmosférica Antonio José Fernández Espinosa, 2001

Principios de química Richard E. Dickerson, 1992-11 Esta tercera edición de Principios de Química, se ha proyectado para ser utilizada en un curso universitario de Química general, que debe servir tanto como una visión de la Química para los especialistas, como de una buena base para los estudios posteriores de las disciplinas de Química. Por consiguiente, hay varios capítulos que pretenden introducir las diferentes áreas de la Química, incluida la inorgánica, nuclear, orgánica y

Bioquímica, y se intenta, a lo largo de todo el libro, colocar a la Química en su marco histórico y cultural. Al mismo tiempo, se presentan los aspectos cuantitativos de la Química de forma consecuente con su importancia, de manera que resulte fácil apoyarse en ellos en los cursos posteriores.

introduccion a la química descriptiva ,

Química ,2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Química general Gordon M. Barrow,1975-11 Sin introducir grandes modificaciones en los temas que habitualmente se encuentran en un texto de esta clase, el autor ha tratado de mostrar que la Química se origina para proporcionar una perspectiva adecuada y entrar en el tema.

Química G. William Daub,William S. Seese,1996

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Química para el ingreso ,

Química. Fundamentos experimentales Robert W. Parry,1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseanza donde se cursan estudios de nivel medio y preuniversitario.Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día.Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

Principios de Química. La búsqueda del conocimiento. P. W. Atkins,Loretta Jones,2006

Introduccion a la Quimica ,

Introducción a la química ambiental Stanley E. Manahan,2006 Este nuevo libro de S. Manahan es una introducción concisa a la Química ambiental que está estrechamente relacionada con sus textos sobre Química verde y Ecología industrial. El concepto de Química ambiental va más allá de la mera discusión sobre la contaminación y los problemas ambientales y pretende enfocar los conocimientos actuales en la resolución de esos problemas basándose en la sostenibilidad y en la prevención de la contaminación. De esta forma, se invocan constantemente disciplinas emergentes como la Ecología industrial, la Química verde y la Ingeniería verde.

Química. La ciencia básica DOMINGUEZ REBOIRAS, MIGUEL ANGEL, 2006-01-01 Esta obra se ha escrito pensando en las necesidades de los estudiantes de los primeros cursos de Facultades de Ciencias, Farmacia, Ciencias Ambientales y de la Salud, y Escuelas Técnicas.

Química: teoría y problemas José María Teijón, José Antonio García Pérez, 1996 Explica cada uno de los conceptos de la Química General y los aplica en más de 500 problemas resueltos. Incluye prácticas de laboratorio y tests de autoevaluación. También adecuado para COU-LOGSE.

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