

# S By Pavel Tsatsouline

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 web mar 27 2008 table 2 determining the atomic mass of

beanium isotopes and molar mass of beanium bean type  
 number of beans mass of beans of this kind atomic mass of  
 this bean isotope abundanc e kidney bean 32 36 1 122 black  
 eye bean 98 23 0 25 white bean 30 16 0 50 calculation of  
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 totalmass atoms total mass isotope 1 atoms 1 avg mass 1  
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 atoms 3 avg mass 3 total mass ofallatoms total of atoms

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 mass unit mass of one proton mass of one neutron a sample  
 of cesium cs has the following abundance cs 132 20 0 cs 133  
 75 3 cs 134 4 7 weighted average atomic mass  $132 \times 0.200$   
 $133 \times 0.752$   $134 \times 0.047$  132 85 amu

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 isotope what does it mean to say that the atoms in a sample  
 of an element are isotopes of each other ordinary beans are a  
 lot bigger than atoms but perhaps they can give you one or  
 two clues about isotopes we have talked in class about  
 isotopes atoms of the same element that differ in

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 average atomic mass write a mathematical equation that  
 shows how you would determine the average atomic mass of

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 units total all beans 100 beanium isotope blackium beanium  
 isotope brownium beanium

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web beanium isotope b brownium beanium isotope c  
 whiteium of atoms beans mass beans in cup mass empty cup  
 mass of atoms beans 2 minus 3 average mass of 1 bean 4  
 divided by 1 of beans 1 a b c x 100 section 3 calculate the  
 average atomic mass of beanium

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 exercise is designed to show you isotopes of an element in a  
 simulation form you will be asked to gather data about the  
 isotopes and organize the data if atoms were as large as  
 beans they could be sorted counted and massed

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web in the beanium lab students obtained a sample of  
 beanium a fictitious element in order to learn about isotopes  
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 brown and speckled which have different masses and  
 different percent abundances

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 this experiment click the card to flip calculate from  
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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web procedure obtain a sample of the classroom mixture of beanium sort your beanium into its three isotopic bean types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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web terms in this set 14 what are the objective of the lab calculate from experimental data the atomic mass of the fictitious element beanium by using a weighted average and simple average decide which method gives a better representation of the atomic mass of beanium

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web good luck saftey 1 do not eat the beans 2 do not throw

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