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1  $\text{Fe} + 2 \text{SO}_4 \rightarrow 3 \text{H}_2\text{O} + 2 \text{C}_2\text{H}_6\text{O}_3$  1  $\text{H}_2\text{O} + 2 \text{C}_2\text{H}_6\text{O}_2$

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$\text{Fe}_2\text{O}_3$  b  $\text{H}_2 + \text{Cl}_2 \rightarrow 2 \text{HCl}$  c  $2 \text{Ag} + \text{H}_2\text{S} + \text{Ag}_2\text{S} \rightarrow \text{H}_2$  d  $\text{CH}_4 + 2 \text{O}_2 \rightarrow \text{CO}_2 + 2 \text{H}_2\text{O}$

e  $2 \text{H}_2\text{O} + 2 \text{Hg} + \text{O}_2 \rightarrow 2 \text{CO} + 3 \text{H}_2\text{O} + \text{CO}_2$  3  $\text{H}_2 + 3$  write balanced

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teams 5 teams 6 teams 7 teams 8 teams 9 teams 10 teams

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