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abundanc e kidney bean 32 36 1 122
black eye bean 98 23 0 25 white bean 30 16 0 50 calculation of atomic mass of beanium 36 32 1 amu
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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass 132 x 0 200 133 x 0 752 134 x 0 047 132 85 amu

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c x 100 section 3 calculate the average atomic mass of beanium

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show you isotopes of an element in a simulation form you will be asked to gather data about the isotopes and organize the data if atoms were as large as beans they could be sorted counted and massed

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by

counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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