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used in an acid base experiment in
order to determine the concentrations
of solutions of acids and bases through
the titration process we are able to
identify physical changes to the mixture
such as the colour change to indicate
the end point of the experiment
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are usually used to find the amount of a
known acidic or basic substance
through acid base reactions the analyte
titrand is the solution with an unknown
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solution with a known molarity that will
react with the analyte

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web conclusion the purpose of the
experiment was to identify the
unknown solid acid 17 by calculating its
molar mass this was done by titration of
the acid with sodium hydroxide in order
to find the volume of sodium hydroxide
required to reach the end point and
that value was used in unit analysis to
find the molar mass and identify the

unknown solid acid
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web by utilizing the half equivalence
point on the titration curve the pka of
the acid analyte was determined to be 3
655 furthermore using the
stoichiometric ratio between the naoh
and the acid analyte 1 1 we could
experimentally determine the number
of moles of the acid in the solution to be
000731 moles making the experimental
molar mass

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experiment 10 11 part 1 acid base
titration abstract the purpose of this
experiment is to observe the titration of
hydrochloric acid a strong acid with
sodium hydroxide a strong base and
acetic acid a

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web goals understand that parts of an
acid base titration be able to determine
the K_a or K_b from ph data associated
with the titration of a weak acid or base
be able to determine the molar mass of
a solid monoprotic acid from titration

data be able to calculate K_{a1} and K_{a2}
for a polyprotic acid by the end of this
lab students should be able to
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neutralization of hydrochloric acid by
sodium hydroxide the mole ratio of acid
to base is 1 1 $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{O l}$
one mole of hcl hcl would be fully
neutralized by one mole of naoh naoh
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unknown solid acid the purpose of this
experiment was to determine the
standardization of sodium hydroxide
naoh and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and
the molar mass of an unknown solid
acid using the molar mass of the
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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{h}_2\text{c}_2\text{o}_4$ aq 2naoh aq $\text{na}_2\text{c}_2\text{o}_4$ $4\text{h}_2\text{o}$ to differentiate an acid from a base in a

solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to

as the equivalence point considering the 1 1 stoichiometry of this acid base reaction naoh aq c_6h_4 cooh cook aq c_6h_4 coona cook aq h_2o l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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aug 15 2022 leon molla është një
qytetar shqiptar i cili u shpall non grata
nga turqia dhe u dënua me dy vite
burgim nga gjykata e lezhës pasi ishte
organizatori i një proteste në vitin

**faquir 73 ala delta serie roja by
daniel monedero alonso** - Mar 15
2022

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