

# Oh Happy Day Song Sister Act

Mike Jess

What is the formula of "magnesium hydroxide"? | Socratic Feb 27, 2017 ·  $\text{Mg}(\text{OH})_2$  Now we know that hydroxides are salts of  $\text{HO}^-$ , and some metal ion. Now if the parent metal has an electronic configuration of 2:8:2, then there are 12 electrons, ...

**Question #627ef - Socratic** functional groups like -OH, -CHO, -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO, or -OH and -CO-. Aldose shows different reactions ... *What is the product of the following reaction?* 1)  $\text{CH}_3\text{OH}$  - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write....  $\text{NH}_4^+ + \text{HO}^- \rightleftharpoons \text{NH}_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$  For methanol, the acid base reaction would proceed ... What are the spectator ions in the reaction between sodium Oct 16, 2016 · The nitrate and the sodium ions.  $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightleftharpoons \text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$  The net ionic equation is:  $2\text{Ag}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightleftharpoons \dots$  *Could a buffered solution be made by mixing aqueous ...* - Socratic Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ... *If the pH=9 for a solution of KOH(aq), what is the ...* - Socratic Aug 18, 2016 · If the pH = 9 for a solution of KOH (aq), what is the concentration of  $\text{H}^+$ ? **What makes for a good leaving group? + Example - Socratic** Jun 23, 2015 · A good leaving group has to be able to part with its electrons easily enough, so typically, it must be a strong acid or weak base relative to other substituents on the same ... What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a  $\text{M}^+$  ion. Hydroxide anion,  $\text{OH}^-$ , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ... Question #44ec4 - Socratic That's a silly way to say carbon dioxide!  $\text{CO}_2(\text{g}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq})$  May be wrong, but it seems reasonable. The gas would interact ... **0.10 moles of sodium cyanate dissolved in 250cm<sup>3</sup> distilled** Dec 28, 2015 · The cyanate anions will react with water to form isocyanic acid, HNCO, and hydroxide anions,  $\text{OH}^-$ . This tells you that you can expect the pH of the resulting solution to be ...

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