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Terry C. Jones

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p4o10 pf5 scl2 sulfur dichloride n2o4 dinitrogen tetroxide

the a of tetra is dropped before the o of oxide p4o10

tetraphosphorous decoxide the a of deca is dropped before

the o of oxide pf5 phosphorous pentafluoride

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h⁺ the h⁺ in is what defines the acidic properties of a

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contain sulfur phosphorus and nitrogen each pair has one

acid that ends in ic and another that ends in ous these

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in model 2 describe the role of the brønsted lowry base in the

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