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Zhe Xu

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produced step one figure out how many moles of the limiting reagent you have 0.216232 grams mole 21.3 grams 0.66 moles 32 g/mol

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web 7. Na₂C₂O₄ is the limiting reagent percent yield 86.6% 8. only four molecules can be made 9. this amount cannot be weighted by ordinary balances and is worthless 10. nitrogen is the limiting reagent 11. yes methane is the limiting reagent 12. C is the limiting reagent 4.33 g of H₂ are left over 13.

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