

Voice In The Wind

M Mosston

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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others

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example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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polyatomic practice 1 name or write the formula for the following polyatomic ions sulfate nitrite perphosphate hypoiodite chlorite phosphite percarbonate bromate hyposulfite permanganate carbonite CO_3^{2-} MnO_3^{1-} SO_5^{2-} BrO_2^{1-} CO_4^{2-} PO_5^{3-} ClO_1 IO_2^{1-} PO_4^{3-} NO_2^{1-} SO_4^{2-}

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web polyatomic ions are charged groups of atoms an example is ammonium ion NH_4^+ it has five atoms one nitrogen and four hydrogens that

share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is NH_4Cl and ammonium sulfide is NH_4S

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion NO_3^- contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit
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