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but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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particular element by the number of positively charged massive nuclear particles i e z 1 we got hydrogen z 2 we got helium z 3 lithium z 47 we got silver the nucleus can also contain neutrons massive nuclear particles of zero charge

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