

# Tears Are Not A Sign Of Weakness

**Terry C. Jones**

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 relationship between heat and  $\Delta h_{rxn}$  is  $\Delta h_{rxn} q_{rxn} q_{calorimeter} mcs\Delta t$   
 $7 3 14 7 3 14 \Delta h_{rxn} q_{rxn} q_{calorimeter} mcs\Delta t$

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 ml of 1 00 m naoh at 25 0 c in a foam cup calorimeter a  
 reaction occurs and the temperature rises to 32 0 c calculate  
 the enthalpy change in kj for this reaction assume the  
 densities of the solutions are 1 00 g ml and that

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 answer the initial temperature of the copper

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 reaction as written 2 nh 3 g 3 o 2 g 2 ch

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 than to t 2 a the final temperature is closer to t 1 than to t 2

the final temperature is exactly halfway between  $t_1$  and  $t_2$   
 the final temperature is exactly halfway between  $t_1$  and  $t_2$   
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