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web 1 $3\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{HBr}$ type
 of reaction double displacement 2 $3\text{CaOH} + 2\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{AlOH}_3$ type of
 reaction double displacement 3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single
 displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$
 type of reaction combustion 5
 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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web a chemical reaction is a process in which one or more substances are converted to one or more different substances the starting substances are

called the reactants and the new substances that form are called the products

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leaves turning color fireworks food spoiling 2 differentiate between a physical and chemical change physical does not change the identity of the matter can be changed back easily chemical new substance is created with new physical and chemical